

## Chapter 4 Atomic Structure Workbook Answers

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### Chapter 4 Atomic Structure Workbook

Chapter 4 Atomic Structure Section 4.1 Studying Atoms (pages 100-105) This section discusses the development of atomic models. Reading Strategy (page 100) Summarizing As you read, complete the table about atomic models. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of

### Chapter 4: Atomic Structure Section 4.1: Studying Atoms

Physical Science Reading and Study Workbook Level B Chapter 4 33 IPLS Chapter 4 Atomic Structure Summary 4.1 Studying Atoms • The ancient Greek philosopher Democritus believed that all matter consisted of extremely small particles that could not be divided. • Aristotle did not think there was a limit to the number of times matter could be ...

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Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these three

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104 Chapter 4 • The Structure of the Atom ■ Figure 4.2 In his book A New System of Chemical Philosophy, John Dalton presented his symbols for the elements known at that time and their possible combinations.

### Chapter 4: The Structure of the Atom

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### Section 3 Modern Atomic Theory Workbook Answers

Chapter 4: Atomic Structure Vocabulary Review. the sum of the protons and neutrons in the nucleus of that atom. atoms of the same element that have different numbers of neutrons and different mass numbers. This activity was created by a Quia Web subscriber.

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